

|   | U                        | <del>1</del>                        | Document ID      | Issue Date | Pages |
|---|--------------------------|-------------------------------------|------------------|------------|-------|
| 1 | <input type="checkbox"/> | <input checked="" type="checkbox"/> | US 5958367<br>A  | 19990928   | 52    |
| 2 | <input type="checkbox"/> | <input checked="" type="checkbox"/> | WO 9818884<br>A2 | 19980507   | 34    |
| 3 | <input type="checkbox"/> | <input checked="" type="checkbox"/> | WO 9815500<br>A  | 19980505   | 120   |
| 4 | <input type="checkbox"/> | <input checked="" type="checkbox"/> | WO 9631434<br>A  | 19961010   | 53    |

|   | Title  | Current<br>OR | Current<br>XRef   |
|---|--|---------------|---|
| 1 | Methods for preparing porous metal oxides  | 423/701       | 423/702;<br>423/703;<br>423/704;<br>423/705;<br>423/706;<br>423/707;<br>423/708;<br>423/713 |
| 2 | NANOSTRUCTURED AQUEOUS FUELS   |               |   |
| 3 | Stable, hexagonally packed, mesoporous metal oxide molecular sieves - have a well defined structure, are resistant to pore collapse on removal of the templating molecule and are thermally stable |               |   |
| 4 | Prodn. of hexagonally packed mesoporous metal oxide(s) for e.g. catalysts - where the mesostructures are resistant to pore collapse after removal of surfactants and are thermally stable          |               |   |

|   | Retrieval<br>1<br>Classif | Inventor                  | S                                   | C                        | P                        | 2                        | 3                        | 4                        | 5                        |
|---|---------------------------|---------------------------|-------------------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|--------------------------|
| 1 |                           | Ying, Jackie Y.<br>et al. | <input checked="" type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> |
| 2 |                           | YING, JACKIE Y et<br>al.  | <input checked="" type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> |
| 3 |                           | ANTONELLI, D M et<br>al.  | <input checked="" type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> |
| 4 |                           | ANTONELLI, D M et<br>al.  | <input checked="" type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> | <input type="checkbox"/> |

|   | Image Doc.<br>Displayed | PT                       |
|---|-------------------------|--------------------------|
| 1 | US 5958367              | <input type="checkbox"/> |
| 2 | WO 9818884 A2           | <input type="checkbox"/> |
| 3 | WO 9815500 A1           | <input type="checkbox"/> |
| 4 | WO 9631434 A1           | <input type="checkbox"/> |

=> d

L1 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2002 ACS  
RN 12254-17-0 REGISTRY  
CN Aluminum barium oxide (Al12BaO19) (9CI) (CA INDEX NAME)  
OTHER CA INDEX NAMES:  
CN Aluminate (Al12O192-), barium (1:1)  
CN Aluminum barium oxide (BaAl12O19) (8CI)  
CN Barium aluminate (BaAl12O19) (6CI, 7CI)  
OTHER NAMES:  
CN Barium aluminum oxide (BaAl12O19)  
CN Barium hexaaluminate  
CN Barium hexaaluminate (BaAl12O19)  
DR 259686-57-2  
MF Al . Ba . O  
AF Al12 Ba O19  
CI TIS  
LC STN Files: CA, CAOLD, CAPLUS, CEN, IFICDB, IFIPAT, IFIUDB, PROMT,  
TOXCENTER, USPAT2, USPATFULL

| Component | Ratio | Component<br>Registry Number |
|-----------|-------|------------------------------|
| O         | 19    | 17778-80-2                   |
| Ba        | 1     | 7440-39-3                    |
| Al        | 12    | 7429-90-5                    |

265 REFERENCES IN FILE CA (1962 TO DATE)  
25 REFERENCES TO NON-SPECIFIC DERIVATIVES IN FILE CA  
266 REFERENCES IN FILE CAPLUS (1962 TO DATE)  
11 REFERENCES IN FILE CAOLD (PRIOR TO 1967)

=> d his

(FILE 'HOME' ENTERED AT 12:59:12 ON 20 DEC 2002)

FILE 'REGISTRY' ENTERED AT 12:59:19 ON 20 DEC 2002  
L1 1 S BARIUM HEXAALUMINATE

=> s l1

L2 265 L1

=> s nanometer# or millimicron# or micron#

14775 NANOMETER#

107 MILLIMICRON#

14261 MICRON#

L3 28784 NANOMETER# OR MILLIMICRON# OR MICRON#

=> s nm

L4 430888 NM

=> d 1

L4 ANSWER 1 OF 430888 CA COPYRIGHT 2002 ACS  
AN 137:392677 CA  
TI Design and performance of a microchip electrophoresis instrument with  
sensitive variable-wavelength fluorescence detection  
AU Belder, Detlev; Deege, Alfred; Maass, Martin; Ludwig, Martin  
CS Abteilung fur Chromatographie, Max-Planck-Institut fur Kohlenforschung,

Mulheim an der Ruhr, D-45470, Germany  
SO Electrophoresis (2002), 23(14), 2355-2361  
CODEN: ELCTDN; ISSN: 0173-0835  
PB Wiley-VCH Verlag GmbH  
DT Journal  
LA English

RE.CNT 30 THERE ARE 30 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

=> s 12 and 13

L5 2 L2 AND L3

=> d bib,ab 1-2

L5 ANSWER 1 OF 2 CA COPYRIGHT 2002 ACS  
AN 137:81046 CA  
TI Synthesis of **nanometer**-sized particles by reverse micelle  
mediated techniques  
IN Ying, Jackie Y.; Zarur, Andrey  
PA Massachusetts Institute of Technology, USA  
SO U.S., 20 pp.  
CODEN: USXXAM  
DT Patent  
LA English  
FAN.CNT 1

|      | PATENT NO.     | KIND | DATE     | APPLICATION NO. | DATE     |
|------|----------------|------|----------|-----------------|----------|
| PI   | US 6413489     | B1   | 20020702 | US 1998-60733   | 19980415 |
|      | US 2002110519  | A1   | 20020815 | US 2001-993355  | 20011114 |
| PRAI | US 1997-43321P | P    | 19970415 |                 |          |
|      | US 1998-60733  | A1   | 19980415 |                 |          |

AB The present invention relates to a method of producing particles having a particle size of <100 nm and surface areas of at least 20 m<sup>2</sup>/g where the particles are free from agglomeration. The method involves synthesizing the particles within an emulsion having a 1-40% water content to form reverse micelles. In particular, the particles formed are metal oxide particles. The particles can be used to oxidize hydrocarbons, particularly methane.

RE.CNT 34 THERE ARE 34 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L5 ANSWER 2 OF 2 CA COPYRIGHT 2002 ACS  
AN 121:137300 CA  
TI preparation of heat-resistant BaAl<sub>12</sub>O<sub>19</sub> combustion catalyst by solid-state reaction combined with sub-**micron** grinding  
AU Imamura, S.; Ishida, S.; Ebata, E.; Tsurumi, K.; Nishikawa, T.; Tanaka, K.; Koshiga, I.  
CS Dep. Chem., Kyoto Inst. Technology, Kyoto, 606, Japan  
SO Reaction Kinetics and Catalysis Letters (1994), 52(1), 19-26  
CODEN: RKCLAU; ISSN: 0304-4122  
DT Journal  
LA English  
AB Heat-resistant barium hexaaluminate combustion catalyst was prepd. by a conventional solid-state reaction combined with sub-**micron** grinding. The barium hexaaluminate thus prepd. retained almost the same high surface area at high temps. as the one prepd. by the alkoxide method, exhibiting high activity in the catalytic combustion of methane.

=> s 12 and 14  
L6 17 L2 AND L4

=> s 16 not 15  
L7 16 L6 NOT L5

=> d bib,ab

L7 ANSWER 1 OF 16 CA COPYRIGHT 2002 ACS  
AN 136:254318 CA  
TI Vacuum UV-excited composite phosphor emitting persistent-luminescence  
IN Arai, Kiyotaka; Tateiwa, Toshiaki; Oki, Yoshiko; Watanabe, Mie  
PA Nichia Chemical Industries Co., Ltd., Japan  
SO Jpn. Kokai Tokkyo Koho, 8 pp.  
CODEN: JKXXAF  
DT Patent  
LA Japanese  
FAN.CNT 1

|      | PATENT NO.     | KIND | DATE     | APPLICATION NO. | DATE     |
|------|----------------|------|----------|-----------------|----------|
| PI   | JP 2002080843  | A2   | 20020322 | JP 2000-312058  | 20001012 |
| PRAI | JP 2000-199439 | A    | 20000630 |                 |          |

AB The composite phosphor comprises 100 wt. parts of first phosphor grain coated with 0.5-100 wt. parts of second phosphor having sp. surface area 3-50 m<sup>2</sup>/g, wherein the second phosphor emits light having a peak at 200-450 nm wave length upon excitation by vacuum UV, and the first phosphor emits visible ray upon excitation by vacuum UV or light emitted from the second phosphor. The structure inhibits time-course deterioration of luminance of the first phosphor due to the second phosphor coating, and the phosphor composite is suitable for use in plasma display panels and high load fluorescent lamps, e.g., rare gas elec. discharge lamps.

=> d bib,ab 2-17

L7 ANSWER 2 OF 16 CA COPYRIGHT 2002 ACS  
AN 133:108734 CA  
TI Synthesis of barium hexaaluminate phosphors using combinatorial chemistry  
AU Park, Eung Suk; Choi, Yoon Young; Sohn, Kee-Sun; Kim, Chang Hae; Park, Hee Dong  
CS Display Phosphor Group, Korea Research Institute of Chemical Technology, Teajon, 305-600, S. Korea  
SO Han'guk Seramik Hakhoechi (2000), 37(2), 134-139  
CODEN: HSHAF7  
PB Korean Ceramic Society  
DT Journal  
LA Korean

AB The main objective of the present investigation is to show the feasibility of combinatorial chem. by applying this method to phosphor syntheses. In this respect barium hexaaluminate phosphor was prepd. by the split-pool combinatorial method, which enabled much more rapid search of optimum compns. of target phosphors than conventional synthetic methods. Barium hexaaluminate phosphors doped with Eu<sup>2+</sup> exhibit blue emission while those co-doped with Mn<sup>2+</sup> and Eu exhibit green emission. Basically, the phosphor doped with 1.3 mol of Ba and 0.06-0.15 mol of Eu<sup>2+</sup> exhibit the max. value of emission intensity at 435 nm. Under the UV and VUV excitations, the barium hexaaluminate phosphor co-doped with Mn<sup>2+</sup> and Eu<sup>2+</sup> shows strong green emission.

L7 ANSWER 3 OF 16 CA COPYRIGHT 2002 ACS  
AN 132:340695 CA  
TI Luminance saturation properties of PDP phosphors  
AU Okazaki, C.; Shiiki, M.; Suzuki, T.; Suzuki, K.  
CS Hitachi Central Research Laboratory, Kokubunji, Tokyo, Japan  
SO Journal of Luminescence (2000), 87-89, 1280-1282  
CODEN: JLUMA8; ISSN: 0022-2313  
PB Elsevier Science B.V.  
DT Journal  
LA English

AB The authors studied the luminance satn. properties of 5 types of plasma display panel (PDP) phosphors under excitation by an ArF laser (wavelength: 193 nm, pulse width: 25 ns, and frequency: 10 Hz). The relation between luminance and excitation energy d. shows that all the phosphors exhibit luminance satn. above an excitation energy d. level of 0.2 mJ/cm<sup>2</sup>/pulse.

RE.CNT 5 THERE ARE 5 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L7 ANSWER 4 OF 16 CA COPYRIGHT 2002 ACS  
AN 132:327323 CA  
TI Influence of flux on the structure and luminescence of the phosphor BaAl<sub>12</sub>O<sub>19</sub>:Mn  
AU Hong, Guangyan; Zeng, Xiaoqing; You, Hongpeng; Kim, Chang-hong; Pyun, Chong-hong; Park, Cheol-hee; Yu, Byung-yong; Bal, Hyun-sook; Kwon, Ii-fook  
CS Changchun Inst. Applied Chem., Chinese Acad. Sci., Changchun, 130022, Peop. Rep. China  
SO Faguang Xuebao (1999), 20(4), 311-315  
CODEN: FAXUEW; ISSN: 1000-7032  
PB Kexue Chubanshe  
DT Journal  
LA Chinese

AB The phosphor BaAl<sub>12</sub>O<sub>19</sub>:Mn was synthesized by solid state reaction at 1300.degree. under the existence of flux. Its XRD diagram showed the intensity increased with the presence of the flux, the intensity of different crystal planes was changed with the flux, for example, crystal planes (102), (107), (114) and (205) became stronger, while crystal planes (0010) and (304) became weak. The flux not only helped to crystallize the host but also affected on growth of different crystal planes. The UV excitation spectrum consists of 3 bands peaking at 279 nm, 360.5 nm, 384.6 nm, which are due to the 6Al.fwdarw.4A<sub>2</sub>(4F), 6Al.fwdarw.4E(4D), 6Al.fwdarw.4T<sub>2</sub>(4D) of Mn<sup>2+</sup> transitions, resp. Its photoluminescence spectra showed there is a stronger emission band peaking at 514 nm and a weak emission band peaking at 450 nm. The former band is assigned to Mn<sup>2+</sup>-emission, and the latter band may be origin from the host. The emission of Mn<sup>2+</sup> ions indicated that Mn<sup>2+</sup> ions occupy crystallog. site of Al in tetrahedral. Also the influence of different flux on the luminescent intensity of the phosphor BaAl<sub>12</sub>O<sub>19</sub>:Mn varied: H<sub>3</sub>BO<sub>3</sub> decreased its luminescence, AlF<sub>3</sub> improved a little and BaF<sub>2</sub> improved greatly. The VUV excitation spectrum consists of the bands peaking around 150 nm and 195 nm that correspond to the host absorption and the 3d<sup>5</sup>.fwdarw.3d<sup>4</sup>sl Mn<sup>2+</sup> transition. This result reveals that there is an high efficient energy transfer from the host to the activator. The strong absorption at .apprx.150 nm also indicates that the phosphor BaAl<sub>12</sub>O<sub>19</sub>:Mn can act as a better candidate of PDP phosphors.

L7 ANSWER 5 OF 16 CA COPYRIGHT 2002 ACS  
AN 130:359143 CA



TI Phosphor layer and display device using it  
IN Shiigi, Masatoshi; Okazaki, Choichirou; Furukawa, Tadashi  
PA Hitachi, Ltd., Japan  
SO Jpn. Kokai Tokkyo Koho, 10 pp.  
CODEN: JKXXAF

DT Patent  
LA Japanese

FAN.CNT 1

|    | PATENT NO.  | KIND | DATE     | APPLICATION NO. | DATE     |
|----|---|------|----------|-----------------|----------|
| PI | JP 11131059   | A2   | 19990518 | JP 1997-298151  | 19971030 |
| AB | The layer contains phosphor particles which generates visible light by excitation of UV light with wavelength .ltoreq.200 nm and a material having optical absorption in a certain wavelength (not visible light). The device has the layer and an excitation source for generating UV light with wavelength .ltoreq.200 nm. The device gives good images with high contrast. |      |          |                 |          |

L7 ANSWER 6 OF 16 CA COPYRIGHT 2002 ACS

AN 127:72431 CA

TI Synthesis and properties of Eu2+ activated blue phosphors

AU Ekambaram, S.; Patil, K. C.

CS Dep. Inorganic and Physical Chem., Indian Institute Science, Bangalore, 560 012, India

SO Journal of Alloys and Compounds (1997), 248(1-2), 7-12  
CODEN: JALCEU; ISSN: 0925-8388

PB Elsevier

DT Journal

LA English

AB Blue phosphors Eu2+ activated BaMgAl10O17, BaMg2Al16O27, xBaO.6Al2O3 where x = 0.64-1.8 and LaMgAl10O19 were prepd. by the combustion of corresponding metal nitrates (oxidizer) and carbonyldiurea (DFH)/urea (fuel) redox mixt. at 400/500.degree. within 5 min. The phosphors were characterized by exposure to UV light, powder XRD, fluorescence and ESR spectroscopy. The phosphors showed a characteristic emission band at .lambda. = 435-462 nm when they were excited at 254 nm. With an increase in Ba content in xBaO.6Al2O3 (x = 0.64-1.8) the emission band showed a red shift. Addn. of Mn2+ in Eu2+ doped Ba hexa aluminates and Eu2+ doped LaMgAl10O19 resulted in strong green emission at 515 nm. The fine particle nature of combustion derived phosphors was studied by powder d. (55-82% of theor. value), particle size (5.7-9.5 .mu.m) and BET surface area (5-22 m2 g-1) measurements.

L7 ANSWER 7 OF 16 CA COPYRIGHT 2002 ACS

AN 126:67211 CA

TI Fluorescent lamps, operating methods and liquid-crystal display apparatus

IN Saito, Miho; Nishimura, Kyoshi; Yuasa, Kunio

PA Toshiba Lighting & Technology, Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

|    | PATENT NO.  | KIND | DATE     | APPLICATION NO. | DATE     |
|----|---|------|----------|-----------------|----------|
| PI | JP 08273620   | A2   | 19961018 | JP 1995-75615   | 19950331 |
| AB | A cold cathode tubular lamp comprises: a quartz tube contg. Hg and Xe (<200 Torr); an outer and an inner phosphor layer activated by Hg UV (185 and 254 nm) and UV < 200 nm (Xe 147 and 172 nm) |      |          |                 |          |

), resp.; and means for activating the inner phosphor initially, then increasing the Hg vapor pressure for activating the outer phosphor in leaching to a max. luminescence. The lamp is suitable for use as a backlight in liq. crystal display devices.

L7 ANSWER 8 OF 16 CA COPYRIGHT 2002 ACS  
AN 124:327510 CA  
TI Study on property of vacuum UV phosphors used in color plasma display panels  
AU Gu, Zhiqi; Liang, Yiyong  
CS Display Technology Inst., Hangzhou Univ., Hangzhou, 310028, Peop. Rep. China

SO Gongneng Cailiao (1995), 26(Suppl.), 158-9  
CODEN: GOCAEA; ISSN: 1001-9731

PB Gongneng Cailiao Bianjibu  
DT Journal

LA Chinese

AB The luminescent property, relative luminescent intensity and light decay property of the primary color vacuum UV phosphors (Y,Gd)BO<sub>3</sub>:Eu (R), BaMgAl<sub>14</sub>O<sub>23</sub>:Eu (B), BaAl<sub>12</sub>O<sub>19</sub>:Mn (G) excited by 147 nm were discussed and the coating properties of three kinds of phosphors in device manufg. were compared. The exptl. results showed that the use of the three phosphors could obtained satisfactory display indexes.

L7 ANSWER 9 OF 16 CA COPYRIGHT 2002 ACS  
AN 121:241363 CA

TI color-variable fluorescent lamps  
IN Yuasa, Kunio

PA Toshiba Lighting & Technology, Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.  
CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

|    | PATENT NO.  | KIND | DATE     | APPLICATION NO. | DATE     |
|----|---|------|----------|-----------------|----------|
| PI | JP 06076801   | A2   | 19940318 | JP 1992-230766  | 19920831 |
| AB | The title lamp, contg. Hg and rare gases, comprises a 1st and a 2nd phosphor layer emitting a 1st and a 2nd colored light via the excitations by the 185 and the 254 nm Hg line, resp.; and means for changing the intensity ratio between the 185 and the 254 nm line by changing the pulse-duty ratio or the bulb temp. The lamp typically changes the color continuously between greenish and reddish white. |      |          |                 |          |

L7 ANSWER 10 OF 16 CA COPYRIGHT 2002 ACS  
AN 113:45105 CA

TI Preparation and microstructure of porous hexaaluminate ceramics  
AU Machida, Masato; Sirouzu, Masaki; Eguchi, Koichi; Arai, Hiromichi

CS Grad. Sch. Eng. Sci., Kyushu Univ., Kasuga, 816, Japan

SO Nippon Seramikkusu Kyokai Gakujutsu Ronbunshi (1990), 98(6), 554-60  
CODEN: NSKRE2; ISSN: 0914-5400

DT Journal

LA Japanese

AB Heat-resistant porous ceramics were prepd. by sintering sol-gel-derived hexaaluminate fine powders at 1200-1600.degree.. Although the sintered samples showed high porosities (50%), the N<sub>2</sub> permeability was low because of small pore size (<100 nm). The packing of planar particles of hexaaluminate formed 2 kinds of peaks in their pore-size distribution at <10 nm and 100 nm. Second-stage sintering of the crushed powders significantly enhanced the N<sub>2</sub> permeability. In the

samples after 2-stage sintering, the loose packing of large agglomerates formed macropores ( $>10 \mu\text{m}$ ) besides the voids of primary particles. The macropores are effective for high gas permeation. Consequently, the 2nd-step sintering of hexaaluminate gave a mixed structure of micropores and macropores, which is a possible microstructure for application to filtration catalysts.

L7 ANSWER 11 OF 16 CA COPYRIGHT 2002 ACS

AN 103:150695 CA

TI Fluorescent lamp

PA Toshiba Corp., Japan

SO Jpn. Tokkyo Koho, 5 pp.

CODEN: JAXXAD

DT Patent

LA Japanese

FAN.CNT 1

|    | PATENT NO.   | KIND | DATE     | APPLICATION NO. | DATE     |
|----|--|------|----------|-----------------|----------|
| PI | JP 60024151  | B4   | 19850611 | JP 1976-31359   | 19760324 |
| AB | A fluorescent lamp is obtained by coating the inner walls of a discharge tube with a mixt. of a Eu-activated rare earth oxysulfide phosphor emitting in the red (600-640 nm), a Tb-activated rare earth oxysulfide phosphor emitting in the yellow-green (530-570 nm), and a Tm-activated rare earth oxysulfide phosphor emitting in the blue (430-480 nm) region. The blue-emitting component may be Sr halophosphate:Eu or Ba aluminate:Eu. White light-emitting fluorescent lamps are obtained by mixing red-, green-, and blue-emitting phosphors. When using RE <sub>2</sub> O <sub>3</sub> :Eu [RE = rare earth] as the red-emitting phosphor and RE <sub>2</sub> O <sub>3</sub> :Tb as the green-emitting phosphor, a white light-emitting fluorescent lamp with high color rendition and efficiency is obtained. |      |          |                 |          |

L7 ANSWER 12 OF 16 CA COPYRIGHT 2002 ACS

AN 93:15928 CA

TI Refractive index and optical absorption of barium hexaaluminate BaAl<sub>12</sub>O<sub>19</sub>

AU Enke, K.; Mateika, D.

CS Forschungslab., Philips G.m.b.H., Hamburg, D-2000/54, Fed. Rep. Ger.

SO Journal of Materials Science (1980), 15(4), 1066-7

CODEN: JMTSAS; ISSN: 0022-2461

DT Journal

LA English

AB Optical absorption and n measurements on Ba<sub>1-x</sub>Al<sub>12</sub>+(2/3)xO<sub>19</sub> (x = 0.1-0.2) single crystals showed a very high band gap (E<sub>2</sub>  $\approx$  6.1 eV according to  $\lambda_{\text{g}} \approx 205 \text{ nm}$ ), whereas the n and the dispersion as well are rather low. Birefringence measurements were carried out in the visible region with a 100- $\mu\text{m}$  thick sample using conventional ellipsometric methods by means of a Leitz microscope.

L7 ANSWER 13 OF 16 CA COPYRIGHT 2002 ACS

AN 91:82024 CA

TI New tricolor phosphors for gas discharge display

AU Koike, Junro; Kojima, Takehiro; Toyonaga, Ryuya; Kagami, Akiyasu; Hase, Takashi; Inaho, Shuji

CS Tech. Res. Lab., Nippon Hoso Kyokai, Tokyo, Japan

SO Journal of the Electrochemical Society (1979), 126(6), 1008-10

CODEN: JESOAN; ISSN: 0013-4651

DT Journal

LA English

AB The properties of phosphors under vacuum UV excitation were studied to develop tricolor phosphors for use in gas discharge panels to reproduce color TV pictures. The excitation spectra at 100-300 nm and the

radiant efficiency of the phosphors incorporated with exptl. gas discharge cells were detd. Based on these results, (Y,Gd)BO<sub>3</sub>:Eu<sup>3+</sup> (red), BaAl<sub>12</sub>O<sub>19</sub>:Mn (green), and BaMgAl<sub>14</sub>O<sub>23</sub>:Eu<sup>2+</sup> (blue) were selected as the new tricolor phosphors which bring high white luminance and wide color gamut to the color picture display panel.

L7 ANSWER 14 OF 16 CA COPYRIGHT 2002 ACS

AN 81:70539 CA

TI Fluorescence in .beta.-aluminum oxide-like materials of potassium, barium, and lanthanum activated with europium(2+) and manganese(2+) ions

AU Tamatani, Masaaki

CS Toshiba Res. Dev. Cent., Tokyo Shibaura Electr. Co., Ltd., Kawasaki, Japan

SO Japanese Journal of Applied Physics (1974), 13(6), 950-6

CODEN: JJAPA5; ISSN: 0021-4922

DT Journal

LA English

AB In-corporation of Mn<sup>2+</sup> and Eu<sup>2+</sup> ions in .beta.-alumina-like host crystals was studied. Under uv excitation, both BaO.6Al<sub>2</sub>O<sub>3</sub>:Eu, Mn, and La<sub>2</sub>O<sub>3</sub>.11Al<sub>2</sub>O<sub>3</sub>:Eu, Mn show intense green fluorescence caused by the energy transfer from Eu<sup>2+</sup> to Mn<sup>2+</sup>. The efficiency of the luminescence is comparable to that of Zn<sub>2</sub>SiO<sub>4</sub>:Mn when excited by 254 nm light. Emission and excitation spectra of the phosphors were compared with those of KAl<sub>11</sub>O<sub>17</sub>. The difference in the at. packing in the mirror plane of these aluminates appears to be reflected in the emission spectra of Mn<sup>2+</sup>. The quenching temp. for the sensitized fluorescence of Mn<sup>2+</sup> is governed by the Eu<sup>2+</sup> ions, replacing mono-, di-, or tri-valent large cations in these compds.

L7 ANSWER 15 OF 16 CA COPYRIGHT 2002 ACS

AN 81:70533 CA

TI Depreciation of .beta.-aluminum oxide-like phosphors under ultraviolet irradiation

AU Tamatani, Masaaki

CS Toshiba Res. Dev. Cent., Tokyo Shibaura Electric Co., Ltd., Kawasaki, Japan

SO Japanese Journal of Applied Physics (1974), 13(6), 957-65

CODEN: JJAPA5; ISSN: 0021-4922

DT Journal

LA English

AB The effects of uv irradiation on the photoluminescence intensity of the .beta.-alumina-like phosphors were investigated. The 185 nm irradiation produces a broad absorption band, due to color centers, in the uv region. Decrease in the fluorescence intensity of both BaO.6Al<sub>2</sub>O<sub>3</sub>:Eu, Mn, and La<sub>2</sub>O<sub>3</sub>.11Al<sub>2</sub>O<sub>3</sub>:Eu, Mn is attributed to the absorption of part of the excitation energy by the color centers. The irreversible photoionization of Eu<sup>2+</sup> to Eu<sup>3+</sup> ions is responsible for the depreciation of KAl<sub>11</sub>O<sub>17</sub>:Eu without color center formation under the 254 nm irradiation. Most of the depreciation can be interpreted solely in terms of the induced reduction of Eu<sup>2+</sup> absorption bands in the uv region. In KAl<sub>11</sub>O<sub>17</sub>:Eu, Mn, the 254 nm irradiation causes depreciation due to the nonradiative decay of the Mn<sup>2+</sup> excited states, in addition to that due to the photoionization of Eu<sup>2+</sup>. Diffusion of ions and/or vacancies may participate in the irreversible photoionization.

L7 ANSWER 16 OF 16 CA COPYRIGHT 2002 ACS

AN 73:104217 CA

TI Luminescent materials

PA Philips Electronic and Associated Industries Ltd.

SO Brit., 6 pp.

CODEN: BRXXAA

DT Patent  
LA English  
FAN.CNT 1

|      | PATENT NO.  | KIND | DATE     | APPLICATION NO. | DATE |
|------|---|------|----------|-----------------|------|
| PI   | GB 1190520  |      | 19700506 |                 |      |
| PRAI | NL  |      | 19671122 |                 |      |
| AB   | For many photochem. document-copying systems, a paper is required which is sensitive to the transmitted, or reflected radiation used. A further requirement is that the sensitivity is to uv rather than visible wavelength radiation. The majority of such systems use a Hg vapor discharge lamp including a luminescent layer provided on a support, the function of which is to convert the low wavelength radiation to 380-440 nm. The patent describes a luminescent material of formula $BaxSryCazEupAl12O19$ where $x + y + z + p = 1$ ; 1 or 2 of the parameters x, y, and z may equal zero and 0.1 .gtoreq. 0.001. Bal-pEupAl12O19 (I) has the highest conversion efficiency, the best temp. dependence and a max. emissivity at the longest wavelength (435 nm). I is satisfactorily excited by the 365- as well as the 254-nm. Hg line. Only slight effects of oxidn. have been noted. TiO2 may be used as a reflecting layer between the support and the luminescent material. Eight examples are given; for each, radiation intensity curves with both wavelength and temp. are drawn. |      |          |                 |      |